Molecular Retrofitting Adapts a Metal–Organic Framework to Extreme Pressure

Eugene A. Kapustin, †,‡§⊥ Seungkyu Lee, †,‡§⊥ Ahmad S. Alshammari, † and Omar M. Yaghi*,†,‡§⊥,⊥

INTRODUCTION

A unique feature of metal–organic frameworks (MOFs) is that they have “pores without walls”, where the internal space is encompassed by multimetallic junctions and organic linkers rather than continuous impenetrable walls as in traditional porous materials. This junction-and-linker arrangement maximally exposes the adsorptive sites and increases their number, leading to ultrahigh surface areas for MOFs. 1–3 Thus, having pores without walls couples the high storage capacity with the facile diffusion of molecules in and out of the pores, making MOFs useful in natural gas storage, separation of gas mixtures, and selective catalysis. 4–7 It is remarkable that, even with the vast openness of MOF structures, they have been shown to be architecturally, thermally, and chemically stable: properties that are essential for their development from basic science to applications and commercialization. 8 As more applications are essential for their development from basic science to architecturally, thermally, and chemically stable: properties that vast openness of MOF structures, they have been shown to be and selective catalysis. 4

ABSTRACT: Despite numerous studies on chemical and thermal stability of metal–organic frameworks (MOFs), mechanical stability remains largely undeveloped. To date, no strategy exists to control the mechanical deformation of MOFs under ultrahigh pressure. Here, we show that the mechanically unstable MOF-520 can be retrofitted by precise placement of a rigid 4,4′-biphenyldicarboxylate (BPDC) linker as a “girder” to afford a mechanically robust framework: MOF-520-BPDC. This retrofitting alters how the structure deforms under ultrahigh pressure and thus leads to a drastic enhancement of its mechanical robustness. While in the parent MOF-520 the pressure transmitting medium molecules diffuse into the pore and expand the structure from the inside upon compression, the girder in the new retrofitted MOF-520-BPDC prevents the framework from expansion by linking two adjacent secondary building units together. As a result, the modified MOF is stable under hydrostatic compression in a diamond-anvil cell up to 5.5 gigapascal. The increased mechanical stability of MOF-520-BPDC prohibits the typical amorphization observed for MOFs in this pressure range. Direct correlation between the orientation of these girders within the framework and its linear strain was estimated, providing new insights for the design of MOFs with optimized mechanical properties.

MOF crystals provides means for identifying the weakest component of the framework. We also introduce the concept of molecular retrofitting and demonstrate its use in adapting such a framework to extreme mechanical stress (Figure 1). Specifically, we examined the mechanical response of porous MOF-520 crystal structure, Al₂(μ-OH)₆(HCOO)₄(BTB)₄ (BTB = 1,3,5-benzenetribenzoate), to hydrostatic pressures from ambient up to 3 gigapascal (GPa). 5 We observed expansion of the framework along two crystallographic axes as a function of pressure, and thereby identified the most vulnerable element of this MOF. We proceeded to introduce 4,4′-biphenyldicarboxylate (BPDC) linkers as molecular “girders” of ideal size and shape to covalently fit into the backbone of MOF-520. In this way, the original fragile MOF-520 was retrofitted by BPDC to become a mechanically robust framework as evidenced by the fact that it remains crystalline up to 5.5 GPa and on subsequent decompression.

Typically, the extreme pressure regime is achieved by loading of single-crystalline or powder sample between the two

Received: April 19, 2017
Published: June 7, 2017
minuscule culets of opposing diamonds of a diamond-anvil cell (DAC). A pressure transmitting medium (PTM) uniformly surrounds the sample to provide hydrostaticity. Usually the crystal structures of MOFs under these conditions experience significant distortions, both in the multimetallic SBUs and in the organic linkers, and even undergo reversible/irreversible phase transitions. The possibility of having the pressure medium molecules diffuse into the MOF pores is unique as it allows study and evaluation of framework deformity. In most cases, the penetration of solvent molecules under extreme pressure inside the MOF (termed overhydration effect) creates internal pressure, which expands the already filled MOF. At initial compression, the framework expands, but then contracts at higher pressures. The behavior of the framework under these conditions provides an opportunity to identify its weakest points. However, the overhydration effect usually results in reduction of long-range order in the crystal, and the sample becomes irreversibly amorphous. Typically, this lack of crystallinity under pressure has prohibited the study of framework distortions by means of single-crystal X-ray diffraction techniques. These challenges are addressed in the present study.

RESULTS AND DISCUSSION

Mechanical Damage of MOF-520 under Extreme Pressure. First, we studied the mechanical response of the crystal structure of MOF-520 toward the increase of the hydrostatic pressure in methanol/ethanol pressure transmitting media in a DAC. MOF-520 is composed of Al-based SBUs linked together with BTB linkers. The secondary building units in MOF-520 are octametallic rings with Al octahedra sharing corners through eight −OHs and four formate ligands. Every SBU is linked by 12 BTBs, resulting in a (12,3)-connected net with fön topology (Figure 2). MOF-520 crystallizes in the non-centrosymmetric space group P4221 with unit cell parameters of 18.3754(6) Å (a) and 37.6893(12) Å (c) and a unit cell volume of 12726.0(9) Å³.

The single-crystalline sample of MOF-520 was placed into a DAC with a culet size of 500 μm, a tungsten gasket, and a methanol/ethanol mixture (4:1) as a penetrating PTM at room temperature. Several rubies were also placed next to the MOF crystal to monitor the pressure using the ruby fluorescence method. The sample was immersed in the PTM for solvent exchange for at least a week prior to single-crystal X-ray data collection. High-pressure data was collected at the 12.2.2 high-pressure beamline at the Advanced Light Source synchrotron (LBNL). The crystal structure was solved and refined at 10⁻⁴, 0.15(2), 0.86(2), 1.47(2), 2.24(2), and 2.82(2) GPa upon compression of MOF. At pressures higher than 2.82 GPa, the sample turned amorphous, and the deterioration of the data prevented the structure solution and refinement. Upon further decompression, the data collection was unsuccessful due to complete degradation of the sample: the crystal cracked and disintegrated into several pieces (Figure S9). The crystallographic data and the pore environment information are shown in Table 1. The pore volume and electron count of unassigned electron density within the pore were calculated using the SQUEEZE algorithm.
The bulk strain of MOF-520 crystal structure in the range from $10^{-4}$ to 3 GPa was found to be non-monotonic (Figure 3). Upon initial compression up to 0.86 GPa, the crystal structure of MOF expands by almost 1% due to the overhydration effect. A similar observation was reported before for prototypical frameworks, such as MOF-5, ZIF-8, HKUST-1, and UiO-67.19−22 The amount of unassigned electron density, corresponding to PTM molecules, increases by almost 40% from 9299 electrons after DAC loading to 15064 at 0.15(2) GPa, signaling the penetration of the solvent inside the pore. Despite the uncertainty in the calculation of the electron density by SQUEEZE algorithm caused by the poor quality of the experimental data (low resolution, lack of low-angle reflections, overlap with the reflections from two diamonds), this observation is consistent with the pore volume increase by almost 300 Å³ due to internal pressure. The analysis of anisotropy of compression reveals that the structural expansion along $a$ and $b$ crystallographic axes prevails the contraction along the $c$ direction. Upon further increase of pressure, MOF-520 starts to compress along all directions, resulting in the overall decrease in the unit cell volume. Thus, the relative change in volume reaches $\sim$0.5% at 2.82(2) GPa. The pore shrinks in size monotonically, leading to the “activation” of the porous material: the number of methanol molecules inside the pore at 2.82(2) GPa is even smaller than such at 0.15(2) GPa. The analogous “activation” was previously reported for ZIF-8 under pressure.21

The bulk strain of MOF-520 crystal structure in the range from $10^{-4}$ to 3 GPa was found to be non-monotonic (Figure 3). Upon initial compression up to 0.86 GPa, the crystal structure of MOF expands by almost 1% due to the overhydration effect. A similar observation was reported before for prototypical frameworks, such as MOF-5, ZIF-8, HKUST-1, and UiO-67.19−22 The amount of unassigned electron density, corresponding to PTM molecules, increases by almost 40% from 9299 electrons after DAC loading to 15064 at 0.15(2) GPa, signaling the penetration of the solvent inside the pore. Despite the uncertainty in the calculation of the electron density by SQUEEZE algorithm caused by the poor quality of the experimental data (low resolution, lack of low-angle reflections, overlap with the reflections from two diamonds), this observation is consistent with the pore volume increase by almost 300 Å³ due to internal pressure. The analysis of anisotropy of compression reveals that the structural expansion along $a$ and $b$ crystallographic axes prevails the contraction along the $c$ direction. Upon further increase of pressure, MOF-520 starts to compress along all directions, resulting in the overall decrease in the unit cell volume. Thus, the relative change in volume reaches $\sim$0.5% at 2.82(2) GPa. The pore shrinks in size monotonically, leading to the “activation” of the porous material: the number of methanol molecules inside the pore at 2.82(2) GPa is even smaller than such at 0.15(2) GPa. The analogous “activation” was previously reported for ZIF-8 under pressure.21

Table 1. Crystallographic Data and Pore Metrics of MOF-520 before Retrofitting as a Function of Hydrostatic Pressure

<table>
<thead>
<tr>
<th>pressure/GPa</th>
<th>$a$/Å</th>
<th>$c$/Å</th>
<th>unit cell vol/Å³</th>
<th>pore vol/Å³</th>
<th>electron count/e</th>
<th>electron count per pore vol/e Å⁻³</th>
</tr>
</thead>
<tbody>
<tr>
<td>0.0001</td>
<td>18.920(3)</td>
<td>37.190(7)</td>
<td>13313(5)</td>
<td>9391.5</td>
<td>9172</td>
<td>$\sim$0.98</td>
</tr>
<tr>
<td>0.15(2)</td>
<td>19.070(3)</td>
<td>36.930(7)</td>
<td>13430(5)</td>
<td>9653.7</td>
<td>15064</td>
<td>$\sim$1.56</td>
</tr>
<tr>
<td>0.86(2)</td>
<td>19.396(3)</td>
<td>36.569(7)</td>
<td>13475(4)</td>
<td>9629.5</td>
<td>10558</td>
<td>$\sim$1.10</td>
</tr>
<tr>
<td>1.47(2)</td>
<td>19.182(3)</td>
<td>36.534(7)</td>
<td>13443(4)</td>
<td>9495.8</td>
<td>13890</td>
<td>$\sim$1.46</td>
</tr>
<tr>
<td>2.24(2)</td>
<td>19.130(3)</td>
<td>36.480(7)</td>
<td>13350(4)</td>
<td>9528.4</td>
<td>12753</td>
<td>$\sim$1.33</td>
</tr>
<tr>
<td>2.82(2)</td>
<td>19.070(3)</td>
<td>36.409(7)</td>
<td>13241(5)</td>
<td>9441.1</td>
<td>11252</td>
<td>$\sim$1.19</td>
</tr>
</tbody>
</table>

The analogous “activation” was previously reported for ZIF-8 under pressure.21

Introduction of the 4,4′-Biphenyldicarboxylate (BPDC) Linkers as Girders. The formate ligands at the SBU of MOF-520 were fully replaced with carboxylic functionalities following the protocol of the coordinative alignment (CAL) method.26 The distance between formate ligands of two adjacent SBUs is ca. 9.5 Å in MOF-520, which allows the precise fitting of the 4,4′-biphenyldicarboxylate of comparable length (10.1 Å). Thus, by soaking MOF-520 single crystals in a saturated solution of H₂BPDC, it was possible to

Figure 2. MOF-520 is built from Al-based octametallic secondary building units and organic BTB linkers. Introduction of H₂BPDC by the CAL method into MOF-520 leads to the new retrofitted MOF-520-BPDC possessing new skl topology. BTB organic linkers are reduced to orange links, and Al-based SBU to blue polyhedral. Atom color scheme: C, black; O, red; Al, blue polyhedra. H atoms are omitted for clarity. Yellow balls indicate the space in the octahedral arrangement of building units in the framework.
replace two formates, linking two SBUs by a rigid molecular girder. The loading rate of this new organic linker was monitored by digested $^1$H NMR. After 4 days, the crystals of the resulting MOF-520-BPDC, $\text{Al}_8(\mu_\text{-OH})_8(\text{BPDC})_2(\text{BTB})_4$, were washed with anhydrous $N,N$-dimethylformamide, and the SXRD data were collected at 100 K at ambient pressure. Noteworthy, the introduction of the new linker does not decrease the crystallinity of the sample: the resolution of the data collected at 100 K for MOF-520-BPDC is comparable with such for pristine MOF-520 (Tables S1 and S2). Since the NMR spectra represent the whole batch of the material but not the individual single crystal, during the structure solution and refinement, the occupancy of BPDC girder was estimated by means of SXRD. The sample was fully characterized in terms of surface area, thermal stability, and composition after activation in order to directly compare the MOFs before and after girder installation [for details see the Supporting Information].

The introduction of the 4,4′-biphenyldicarboxylate girder does not change the symmetry of the MOF, and the space group of MOF-520-BPDC remains $P4_22_12$. At the same time, the structure is expanded along the crystallographic axes $a$ and $b$ by 0.8 Å but compressed along the longest axis $c$ by 1 Å (Table 2). This structural modification leads to the overall expansion of the unit cell by more than 700 Å$^3$. As expected, the crystallographic density of the framework is increased after modification by just 0.05 g/cm$^3$ due to the expansion of the unit cell. The pore volume estimated from the nitrogen adsorption measurement at 77 K is decreased by almost 25% from 1.28 cm$^3$/g for MOF-520 to 0.91 cm$^3$/g for MOF-520-BPDC. Similar reduction of the BET surface area was observed:

Figure 3. Relative changes in unit cell parameters of MOF-520 (left) and MOF-520-BPDC (right) with pressure. Filled and open symbols correspond to increasing pressure and decompression data, respectively. The standard deviations of values are smaller than the size of symbols. The two different regimes of compression in MOF-520 are highlighted with yellow and blue, respectively. The ellipsoids of strain in red are calculated for both structures using PASCAl software; only the regime of compression was used for the calculation of ellipsoid for MOF-520.
connected net. While the overall symmetry of the unit cell is the crystallographic axes (Figure 2). In MOF-520-BPDC the girders are directed along skl possesses the new and thus far unreported MOF toward the extreme pressure.27 The single-crystalline introduction of the new linker may increase the rigidity of theoretical calculations, it was hypothesized before that the BPDC linker.

increased by almost 0.6 Å due to the incorporation of the pressure transmitting medium.

BPDC are still accessible for the methanol/ethanol mixture as a of the overall decrease in porosity, the pores in the MOF-520- respectively; attributable to the addition of the girder. Despite almost decrease in unit cell volume at 1 GPa is just 0.3%, which is ff also increases the sti of the framework from the inside. This is not surprising considering that, for expansion of the unit cell, the carbon—carbon bonds of the BPDC girder must be stretched. Further compression of the MOF crystal leads to the reduction of the pore volume by ~500 Å³, and the amount of unassigned electron density related to resident molecules of PTM within the pore increases from 9455 electrons at 0.34(2) GPa to 10360 electrons at 5.33(5) GPa.

The anisotropy of compression of MOF-520-BPDC is shown in Figure 3. One can see that the structure compresses along all crystallographic directions monotonically. The linear strain reaches almost 2% along the a (and b) crystallographic axis and 1.5% along the c axis at 5.3 GPa. The relative volume change at 5.3 GPa is about 5.5%. Based on the collected data, the ellipsoid of strain was calculated, showing the median compressibility coefficients of 3 TPa−1 along all three crystallographic directions.28 These values are almost two times smaller than those of parent MOF-520, indicating the corresponding increase of stiffness due to the placement of the girders.

The reason for the drastic difference of MOFs’ response to mechanical stimuli is the removal of distortion of the unit cell between two SBUs along the shortest axes due to the girder. For example, the distance between carbon atoms of the formate ligands of two neighboring SBUs is 9.470(4) Å at 100 K, 10.25(2) Å after closing the DAC, 10.31(2) Å in MOF-520 at 0.86 GPa (regime of expansion), and 9.64(4) Å at 2.82 GPa (regime of compression). In contrast, in the retrofitted MOF, the distance between carboxylate carbon atoms in the girder only shortens by 0.37(2) Å at 5.33 GPa.

While in previously studied MOFs, such as MOF-5 and HKUST-1, the weakest element of the structures is the metal—linker bonds in the metal—oxide SBU, in the MOF-520 and retrofitted MOF-520 the rotated phenylene units of the BTB linkers are the weak points responsible for the deformation of the framework (Supporting Information, section S5). In other words, the structure of MOF-520 suffers from mechanical instability due to the organic linker rather than the inorganic SBU. This study showed how this framework instability can be turned into robustness by molecular retrofitting of MOF-520 with BPDC.

<table>
<thead>
<tr>
<th>pressure/GPa</th>
<th>a/Å</th>
<th>c/Å</th>
<th>unit cell vol/Å³</th>
<th>pore vol/Å³</th>
<th>electron count/e−</th>
<th>electron count per pore vol/e− Å⁻³</th>
</tr>
</thead>
<tbody>
<tr>
<td>0.0001</td>
<td>19.215(4)</td>
<td>36.779(4)</td>
<td>13580(6)</td>
<td>9104.1</td>
<td>8612</td>
<td>~0.95</td>
</tr>
<tr>
<td>0.32(2)</td>
<td>19.2263(19)</td>
<td>36.708(3)</td>
<td>13569(3)</td>
<td>8992.7</td>
<td>6989</td>
<td>~0.77</td>
</tr>
<tr>
<td>0.64(2)</td>
<td>19.1965(10)</td>
<td>36.699(2)</td>
<td>13523.9(16)</td>
<td>9031.8</td>
<td>9455</td>
<td>~1.04</td>
</tr>
<tr>
<td>1.12(2)</td>
<td>19.185(2)</td>
<td>36.720(2)</td>
<td>13515(2)</td>
<td>9080.1</td>
<td>9464</td>
<td>~1.04</td>
</tr>
<tr>
<td>1.67(2)</td>
<td>19.156(2)</td>
<td>36.710(2)</td>
<td>13471(3)</td>
<td>9054.4</td>
<td>9792</td>
<td>~1.08</td>
</tr>
<tr>
<td>2.26(2)</td>
<td>19.123(2)</td>
<td>36.648(3)</td>
<td>13401(3)</td>
<td>9023.6</td>
<td>10519</td>
<td>~1.16</td>
</tr>
<tr>
<td>2.45(2)</td>
<td>19.083(4)</td>
<td>36.725(7)</td>
<td>13374(6)</td>
<td>8888.5</td>
<td>12365</td>
<td>~1.39</td>
</tr>
<tr>
<td>2.86(2)</td>
<td>19.086(2)</td>
<td>36.593(3)</td>
<td>13330(3)</td>
<td>8942</td>
<td>11046</td>
<td>~1.23</td>
</tr>
<tr>
<td>3.32(5)</td>
<td>19.0287(13)</td>
<td>36.5400(17)</td>
<td>13230.8(19)</td>
<td>8825.8</td>
<td>9673</td>
<td>~1.09</td>
</tr>
<tr>
<td>4.20(5)</td>
<td>18.955(4)</td>
<td>36.464(4)</td>
<td>13101(5)</td>
<td>8712.1</td>
<td>9756</td>
<td>~1.12</td>
</tr>
<tr>
<td>4.71(5)</td>
<td>18.933(3)</td>
<td>36.424(4)</td>
<td>13056(4)</td>
<td>8640.2</td>
<td>9979</td>
<td>~1.15</td>
</tr>
<tr>
<td>5.33(5)</td>
<td>18.857(3)</td>
<td>36.302(7)</td>
<td>12909(4)</td>
<td>8572.3</td>
<td>10360</td>
<td>~1.21</td>
</tr>
</tbody>
</table>

*Measured upon decompression.

3630 and 2548 m²/g for MOF-520 and MOF-520-BPDC, respectively; attributable to the addition of the girder. Despite of the overall decrease in porosity, the pores in the MOF-520-BPDC are still accessible for the methanol/ethanol mixture as a pressure transmitting medium.

In the retrofitted MOF, each SBU is now linked by four BPDCs in addition to 12 BTBs, resulting in a new (16,3)-connected net. While the overall symmetry of the unit cell is preserved during the modification, the retrofitted MOF possesses the new and thus far unreported skl topology (Figure 2). In MOF-520-BPDC the girders are directed along the crystallographic axes a and b, while the c direction remains unchanged. The distance between two adjacent SBUs is increased by almost 0.6 Å due to the incorporation of the BPDC linker.

**Mechanical Robustness and Retention of Crystallinity of MOF-520-BPDC under Extreme Pressure.** Our next step was to study the mechanical response of the new modified framework to variations in hydrostatic pressure. Using theoretical calculations, it was hypothesized before that the introduction of the new linker may increase the rigidity of MOF toward the extreme pressure.27 The single-crystalline sample of MOF-520-BPDC was placed into a DAC filled with a methanol/ethanol pressure medium, and single-crystal X-ray diffraction data were collected up to 5.5 GPa. It is worth noting that we did not collect the data at even higher pressures because of the size of the crystal: the distance between two cutlets of diamonds was almost equal to the longest dimension of the sample, and further compression could result in cracking of the MOF crystal. Additionally, the SXRD data were also collected on the subsequent decompression.

Notably, the crystal structure of the retrofitted MOF experiences only one regime of distortion. The installation of rigid girder removes the overhydration effect upon initial compression but also increases the stiffness of the framework. The relative decrease in unit cell volume at 1 GPa is just 0.3%, which is almost five times smaller than in the case of pristine MOF-520. More importantly, the existence of the girder between two SBUs does not hinder the solvent molecules from diffusion inside the pore. The amount of unassigned electron density within the pore increases from 8612 electrons after closing the DAC to 9455 electrons at 0.64(2) GPa. Despite the penetration of pressure medium into the MOF, the internal pressure created by these molecules is not enough to expand the framework from the inside. This is not surprising considering that, for expansion of the unit cell, the carbon—carbon bonds of the BPDC girder must be stretched. Further compression of the MOF crystal leads to the reduction of the pore volume by ~500 Å³, and the amount of unassigned electron density related to resident molecules of PTM within the pore increases from 9455 electrons at 0.34(2) GPa to 10360 electrons at 5.33(5) GPa.

The anisotropy of compression of MOF-520-BPDC is shown in Figure 3. One can see that the structure compresses along all crystallographic directions monotonically. The linear strain reaches almost 2% along the a (and b) crystallographic axis and 1.5% along the c axis at 5.3 GPa. The relative volume change at 5.3 GPa is about 5.5%. Based on the collected data, the ellipsoid of strain was calculated, showing the median compressibility coefficients of 3 TPa⁻¹ along all three crystallographic directions.28 These values are almost two times smaller than those of parent MOF-520, indicating the corresponding increase of stiffness due to the placement of the girders.

The reason for the drastic difference of MOFs’ response to mechanical stimuli is the removal of distortion of the unit cell between two SBUs along the shortest axes due to the girder. For example, the distance between carbon atoms of the formate ligands of two neighboring SBUs is 9.470(4) Å at 100 K, 10.25(2) Å after closing the DAC, 10.31(2) Å in MOF-520 at 0.86 GPa (regime of expansion), and 9.64(4) Å at 2.82 GPa (regime of compression). In contrast, in the retrofitted MOF, the distance between carboxylate carbon atoms in the girder only shortens by 0.37(2) Å at 5.33 GPa.

While in previously studied MOFs, such as MOF-5 and HKUST-1, the weakest element of the structures is the metal—linker bonds in the metal—oxide SBU, in the MOF-520 and retrofitted MOF-520 the rotated phenylene units of the BTB linkers are the weak points responsible for the deformation of the framework (Supporting Information, section S5). In other words, the structure of MOF-520 suffers from mechanical instability due to the organic linker rather than the inorganic SBU. This study showed how this framework instability can be turned into robustness by molecular retrofitting of MOF-520 with BPDC.
Synthetic details, NMR, adsorption data, IR, TGA, PXRD, high-pressure SXRD data, and geometrical parameters of crystal structures (PDF)
Crystallographic data for Δ-MOF-520 (CIF)
Crystallographic data for Δ-MOF-520-BPDC (CIF)

**AUTHOR INFORMATION**

**Corresponding Author**

*E-mail: yaghi@berkeley.edu.*

**ORCID**

Eugene A. Kapustin: 0000-0003-4095-9729
Omar M. Yaghi: 0000-0002-5611-3325

**Notes**

The authors declare no competing financial interest.

**ACKNOWLEDGMENTS**

Support for the synthesis by BASF SE (Ludwigshafen, Germany, Award No. 20153470) and the characterization of compounds by King Abdullah City for Science and Technology (Center of Excellence for Nanomaterials and Clean Energy Applications, Award No. 036931) and the high-pressure single-crystal X-ray diffraction studies by the Center for Gas Separations Relevant to Clean Energy Technologies, an Energy Frontier Research Center funded by the US Department of Energy, Office of Science, Basic Energy Sciences under Award DE-SC0001015. The Advanced Light Source is supported by the Director, Office of Science, Oﬃce of Basic Energy Sciences, of the U.S. Department of Energy under Contract No. DE-AC02-05CH11231. The high-pressure facilities at the Advanced Light Source are supported by COMPRES, the Consortium for Materials Properties Research in Earth Sciences under NSF Cooperative Agreement EAR 11-57758. We thank Dr. Christine Beavers for the synchrotron X-ray diffraction data acquisition support at the beamline 12.2.2 (Advanced Light Source, Lawrence Berkeley National Laboratory); Dr. Markus Kalmutzki and Christian Diercks for discussions of editing of this manuscript. Data reported in this paper are tabulated in the Supporting Information and archived at the Cambridge Crystallographic Data Centre database under the following reference numbers: 1552341−1552360. We acknowledge the collaboration and valuable input of Prince Dr. Turki bin Saud bin Mohammed Al-Saud (President of KACST).

**REFERENCES**


(26) IUCrJ 2014, 5, 607−614.
